

## 2 FERMENTATION: PRINCIPLES AND TECHNOLOGY

### 2.1. INTRODUCTION

The term “fermentation” is derived from the Latin verb *fervere*, to boil, thus describing the appearance of the action of yeast on extracts of fruit or malted grain. The boiling appearance is due to the production of carbon dioxide bubbles caused by the anaerobic catabolism of the sugars present in the extract. However, fermentation has come to have different meanings to biochemists and to industrial microbiologists. Its biochemical meaning relates to the generation of energy by the catabolism of organic compounds, whereas its meaning in industrial microbiology tends to be much broader.

*Fermentation* is a word that has many meanings for the microbiologist:

- 1 Any process involving the mass culture of microorganisms, either aerobic or anaerobic.
- 2 Any biological process that occurs in the absence of O<sub>2</sub>.
- 3 Food spoilage.
- 4 The production of alcoholic beverages.
- 5 Use of an organic substrate as the electron donor and acceptor.
- 6 Use of an organic substrate as a reductant, and of the same partially degraded organic substrate as an oxidant.
- 7 Growth dependant on substrate-level phosphorylation.

#### 2.1.1 THE RANGE OF FERMENTATION PROCESSES

There are five major groups of commercially important fermentations:

- 1 Those that produce microbial cells (or biomass) as the product.
- 2 Those that produce microbial enzymes.
- 3 Those that produce microbial metabolites.
- 4 Those that produce recombinant proteins.
- 5 Those that modify a compound which is added to the fermentation - the transformation process/biotransformation.

#### 2.1.2 THE COMPONENT PARTS OF A FERMENTATION PROCESS

The environment provided for the growth of the process organism must be controlled during the fermentation such that maximum (and reliable) productivity may be achieved. Regardless of the type of fermentation (with the possible exception of some transformation processes) an established process may be divided into SIX basic component parts:

- 1 The chemical environment of the organism should be such that it supports optimum product formation commensurate with the economics of the process. The formulation of media to be used in culturing the process organism during the development of the inoculum and in the production fermenter.
- 2 The culture should be maintained in a pure state throughout the fermentation therefore sterilization of the medium, fermenters and ancillary equipment is required.
- 3 The production of an active, pure culture in sufficient quantity to inoculate the production vessel.

- 4 The growth of the organism in the production fermenter under optimum conditions for product formation.
- 5 The extraction of the product and its purification.
- 6 The disposal of effluents produced by the process.

## **2.2 MICROBIAL CULTURE TECHNIQUES**

### **2.2.1 NUTRIENTS FOR MICROBIAL CULTURE**

Detailed investigations are required to establish the most suitable medium for an individual fermentation. Most fermentations require liquid media, often referred to as **broth**; although some solid substrate fermentations (SSF) are operated. Fermentation media must satisfy all the nutritional requirements of the microorganism and fulfil the technical objectives of the process. All microorganisms require water, sources of energy, carbon, nitrogen, mineral elements and possibly vitamins plus oxygen if aerobic. The nutrients should be formulated to promote the synthesis of the target product, either cell biomass or a specific metabolite. In most industrial fermentation processes there are several stages where media are required. They may include several inoculum (starter culture) propagation steps, pilot scale fermentations and the main production fermentation. The technical objectives of inoculum propagation and the main fermentation are often very different, which may be reflected in differences in their media formulations. Where biomass or primary metabolites are the target product, the objective is to provide a production medium that allows optimal growth of the microorganism. For secondary metabolite production, such as antibiotics, biosynthesis is not growth related. Therefore, media are designed to provide an initial period of cell growth, followed by conditions optimized for secondary metabolite production. At this point the supply of one or more nutrients (carbon, phosphorous, or nitrogen source) may be limited and rapid growth ceases.

Most fermentations, except those involving solid substrates, require large quantities of water in which the medium is formulated. General media requirements include a carbon source, which in virtually all industrial fermentations, provides both energy and carbon units for biosynthesis, and sources of nitrogen, phosphorous and sulphur. Other minor and trace elements must also be supplied and some microorganisms require added vitamins such as biotin and riboflavin. Aerobic fermentations are dependent on a continuous input of molecular oxygen and even some anaerobic fermentations require initial aeration of the media, e.g., beer fermentations. Usually, media incorporate buffers, or the pH is controlled by acid and alkali additions, and antifoam agents may be required. For some processes, precursor, inducer or inhibitor compounds must be introduced at certain stages of the fermentation.

On a large scale one must use nutrient sources to create a medium which will meet as many of the following criteria:

- 1 Produce the maximum yield of product or biomass per gram substrate used.
- 2 Produce the maximum concentration of product or biomass.
- 3 Permit the maximum rate of product formation.
- 4 Minimum yield of undesired product.
- 5 Consistent in quality and readily available throughout the year.
- 6 Cause minimal problems during media making and sterilization.

- 7 Cause minimal problems during other aspects of the production process, especially aeration and agitation, extraction, purification and waste treatment.

The initial step in media formulation is the examination of the overall process based on the stoichiometry for growth and product formation. Thus for an aerobic fermentation:

Carbon & energy + nitrogen + O<sub>2</sub> + other requirements → biomass + products + CO<sub>2</sub> + H<sub>2</sub>O + heat

This primarily involves consideration of the input of the carbon and nitrogen sources, minerals and oxygen, and their conversion to cell biomass, metabolic products, carbon dioxide, water and heat. From this information it should be possible to calculate the minimum quantities of each element required to produce a certain quantity of biomass or metabolite. Typically, the main elemental formula of microbial cells is approximately C<sub>4</sub>H<sub>7</sub>O<sub>2</sub>N, which on the basis of dry weight is 48% C, 6% H, 32% O and 14% N. The elemental composition of baker's yeast for example is C<sub>3.72</sub>H<sub>6.11</sub>O<sub>1.95</sub>N<sub>0.61</sub>S<sub>0.017</sub>P<sub>0.035</sub>K<sub>0.056</sub>. Elemental composition varies slightly with growth rate, but the range is relatively small compared with interspecies differences, particularly between bacteria and fungi. Ideally, knowledge of the complete elemental composition of a specific industrial microorganism allows further media refinement. This ensures that no element is limiting, unless this is desired for a specific purpose.

Once the elemental requirements of a microorganism have been established, suitable nutrient sources can be incorporated into the media to fulfil these demands. However, it is important to be aware of potential problems that can arise when using certain compounds. For example, those that are rapidly metabolized may repress product formation. To overcome this, intermittent or continuous addition of fresh medium may be carried out to maintain a relatively low concentration that is not repressive. Certain media nutrients or environmental conditions may affect not only the physiology and biochemistry, but also the morphology of the microorganism. In some yeasts the single cells may develop into pseudo-mycelium or flocculate, and filamentous fungi may form pellets. This may or may not be desirable, as such morphological changes can influence product yield and other fermentation properties.

The composition of a fermentation medium may be simple to complex, depending on the particular microorganism and its fermentation. Autotrophic microorganisms require only the simplest of inorganic media (inorganic salts, water, nitrogen source, carbon source is fulfilled by CO<sub>2</sub> or by carbonates) and are capable of synthesizing all the complex organic compounds required to sustain life. Fastidious microorganisms on the other hand lack the ability to synthesize many of their sustenance and growth requirements. They require the presence of many simple to complex preformed nutrients in the medium and must have an organic carbon supply to provide for synthesis of cell substances and release of metabolic energy.

Simple and complex media are further subdivided into two categories: synthetic and crude. In a synthetic medium, all the components are specifically defined and known compounds. Each component is relatively pure and the exact concentrations are known. This type of medium has advantages in certain types of studies – as the components and concentrations are defined, the

concentration of one or several can be varied in order to determine the effect on cell growth and product yield. Individual components may be added or deleted as well. However, these are expensive due to the relatively pure ingredients used and yields derived from these media are relatively low. Crude media usually allows much higher yields. They contain crude or ill defined sources of nutrients and growth factors.

The media adopted also depend on the scale of the fermentation. For small-scale laboratory fermentations pure chemicals are often used in well-defined media. However, this is not always possible, simply due to cost, as media components may account for up to 60-80% of process expenditure. Industrial-scale fermentations primarily use cost-effective complex crude substrates, where many carbon and nitrogen sources are almost indefinable. Most are derived from natural plant and animal materials, often by-products of other industries, with varied and variable composition. The effects of such batch-to-batch variations must be determined. Small-scale trials are usually performed with each new batch of substrate, particularly to examine the impact on product yield and product recovery.

The main factors that affect the final choice of individual raw materials are as follows:

- 1 Cost and availability: ideally, materials should be inexpensive and of consistent quality and year round availability.
- 2 Ease of handling in solid or liquid forms, along with associated transport and storage costs, e.g., requirements for temperature control.
- 3 Sterilization requirements and any potential denaturation problems.
- 4 Formulation, mixing, complexing and viscosity characteristics that may influence agitation, aeration and foaming during fermentation and downstream processing stages.
- 5 The concentration of target product to be attained, its rate of formation and yield per gram of substrate utilized.
- 6 The levels and range of impurities and the potential for generating further undesired products during the process.
- 7 Overall health and safety implications.

The fermentation stage is not the sole factor to consider for the final composition of industrial media. Crude substrates may provide initial cost savings, but their higher levels of impurities could necessitate more costly and complex recovery and purification steps downstream as well as increased waste treatment costs. The physical and chemical properties of the formulated medium can also influence the sterilization operations employed. A medium that is easily sterilized with minimal thermal damage is vitally important. Thermal damage not only reduces the level of specific ingredients but can also produce potentially inhibitory by-products that may also interfere with downstream processing. Other media characteristics can affect product recovery and purification and the ease with which cells are separated from the spent medium.

#### **2.2.1.1 Carbon Sources**

A carbon source is required for all biosynthesis leading to reproduction, product formation and cell maintenance. In most fermentations it also serves as the energy source. Carbon requirements may be determined from the biomass yield coefficient ( $Y$ ), an index of the efficiency of conversion of a substrate into cellular material.

$$Y_{\text{carbon(g/g)}} = \frac{\text{biomass produced (g)}}{\text{carbon substrate utilized (g)}}$$

For commercial fermentations, the determination of yield coefficients for all other nutrients is usually essential. Each may be determined by conducting a series of batch culture experiments where the specific substrate is the only growth limiting media component and all other nutrients are in excess. By varying the initial concentration of the growth limiting substrate and then plotting total growth against substrate concentration for each batch, the growth yield (Y) can be estimated. However, the value obtained relates to a specific set of operating conditions, varying pH, temperature etc. can alter the yield coefficient. Various organisms may exhibit different yield coefficients for the same substrate due primarily to the pathway by which the compound is metabolized. Differences can also be seen within an individual organism. *Saccharomyces cerevisiae* grown on glucose has biomass yield coefficients of 0.56 and 0.12 g/g under aerobic and anaerobic conditions, respectively.

As most carbon substrates also serve as energy sources, the organism's efficiency of both adenosine triphosphate (ATP) generation and its utilization are obviously additional key factors. Often, it is very useful, although rather difficult, to estimate how much ATP is required for growth. However, estimates of  $Y_{\text{ATP}}$  (yield of cells per mole of ATP generated during growth) can be calculated if the metabolism of the organism has been fully elucidated.

Carbohydrates are traditional carbon and energy sources for microbial fermentations, although other sources may be used, such as alcohols, alkanes and organic acids. Animal fats and plant oils may also be incorporated into some media, often as supplements to the main carbon source.

### Molasses

Pure glucose and sucrose are rarely used for industrial-scale fermentations, primarily due to cost. Molasses, a by-product of cane and beet sugar production, is a cheaper and more usual source of sucrose. Molasses are concentrated syrups or mother liquors recovered at any one of several steps in the sugar refining process with different names depending on the step from which it is recovered. **Blackstrap** molasses from sugar cane is normally the cheapest and most used sugar source for industrial fermentation. This material is the residue remaining after most of the sucrose has been crystallized from the plant extract. It is a dark-coloured viscous syrup containing 50-60% (w/v) carbohydrates, primarily sucrose, with 2% nitrogenous substances, along with some vitamins and minerals. Overall composition varies depending upon the plant source, the location of the crop, the climatic conditions under which it was grown and the factory where it was processed. The carbohydrate concentration may be reduced during storage by contaminating microorganisms. **Refinery blackstrap** molasses is a similar product obtained from the recrystallization refining of crude sucrose. High test or invert molasses contains approximately 70-75% sugar. It is produced after whole cane juice is partially inverted to prevent sugar crystallization (partially hydrolyzed to monosaccharides). It is preferable to blackstrap molasses as it has lower levels of non fermentable solids and lower shipping charges (on a concentration basis). However, it is only produced during sugarcane overproduction and availability may be questionable. **Beet** molasses are produced in a similar process as for

sugarcane. However, it may be limiting in biotin for yeast growth and a small amount of cane molasses may need to be added in these fermentations. **Hydrol** molasses, a by-product of maize starch processing primarily contains glucose (60%) and a relatively high salt concentration.

### **Malt Extract**

Aqueous extracts of malted barley can be concentrated to form syrups that are particularly useful carbon sources for the cultivation of filamentous fungi, yeast and actinomycetes. Extract preparation is essentially the same as for malt wort production in beer brewing. The composition of malt extracts varies to some extent, but they usually contain approximately 90% carbohydrate, on a dry weight basis. This comprises 20% hexoses (glucose and small amounts of fructose), 55% disaccharides (mainly maltose and traces of sucrose), along with 10% maltotriose, a trisaccharide. In addition, these products contain a range of branched and unbranched dextrans (15-20%), which may or may not be metabolized, depending upon the microorganism. Malt extracts also contain some vitamins and approximately 5% nitrogenous substances, proteins, peptides and amino acids.

Sterilization of media containing malt extract must be carefully controlled to prevent overheating. The constituent reducing sugars and amino acids are prone to generating Maillard reaction products when heated at low pH. These are brown condensation products resulting from the reaction of amino groups of amines, amino acids and proteins with the carboxyl groups of reducing sugars, ketones and aldehydes. Not only does this cause colour change, but it also results in loss of fermentable materials and some reaction products may inhibit microbial growth.

### **Starch and Dextrans**

These polysaccharides are not as readily utilized as monosaccharides and disaccharides, but can be directly metabolized by amylase-producing microorganisms, particularly filamentous fungi. Their extracellular enzymes hydrolyze the substrate to a mixture of glucose, maltose or maltotriose to produce a sugar spectrum similar to that found in many malt extracts. Maize starch is most widely used, but may also be obtained from other cereal or root crops. To allow use in a wide range of fermentations, the starch is usually converted into sugar syrup, containing mostly glucose. It is first gelatinized and then hydrolyzed by dilute acids or amylolytic enzymes, often microbial glucoamylases that operate at elevated temperatures.

### **Sulphite Waste Liquor**

Sulphite waste liquor is derived from the paper pulping industry after wood for paper manufacture is digested to cellulose pulp. It can be used as a dilute fermentation medium for ethanol production by *S. cerevisiae* and the growth of *Torula utilis* for feed. Waste liquors from coniferous trees contain 2-3% (w/v) sugar, which is a mixture of hexoses (80%) and pentoses (20%). Hexoses include glucose, mannose and galactose, whereas the pentose sugars are mainly xylose and arabinose. The liquors derived from deciduous trees contain mainly pentoses. Usually the liquor requires processing before use as it contains sulphur dioxide or calcium hydroxide or calcium carbonate which need to be stripped or removed by precipitation.

with lime. These liquors also require supplementation with sources of nitrogen and phosphorous.

### **Cellulose**

Cellulose is predominantly found as lignocellulose in plant cell walls, which is composed of three polymers: cellulose, hemicellulose and lignin. Lignocellulose is available from agricultural, forestry, industrial and domestic wastes. Relatively few microorganisms can utilize it directly, as it is difficult to hydrolyze. The cellulose component is in part crystalline, encrusted with lignin and provides little surface area for enzyme attack. At present it is mainly used in solid-substrate fermentations to produce various mushrooms. However, it is potentially a very valuable renewable source of fermentable sugars once hydrolyzed, particularly in the bioconversion to ethanol for fuel use.

### **Whey**

Whey is an aqueous by-product of the dairy industry. The annual worldwide production is over 80 million tonnes, containing over 1 million tonnes of lactose and 0.2 million tonnes of milk protein. This material is expensive to store and transport. Therefore, lactose concentrates are often prepared for later fermentation by evaporation of whey, following removal of milk proteins for use as food supplements. Lactose is generally less useful as a fermentation foodstock than sucrose, as it is metabolized by fewer organisms. *S. cerevisiae*, for example, does not ferment lactose. This disaccharide was formerly used extensively in penicillin fermentations and it is still employed for producing ethanol, single cell protein, lactic acid, xanthan gum, vitamin B<sub>12</sub> and gibberellic acid.

### **Alkanes and Alcohols**

*n*-Alkanes of chain length C<sub>10-20</sub> are readily metabolized by certain microorganisms. Mixtures, rather than a specific compound, are usually most suitable for microbial fermentations. However, their industrial use is dependent upon the prevailing price of petroleum. Methane is utilized as a carbon source by a few microorganisms, but its conversion product methanol is often preferred for industrial fermentations as it presents fewer technical problems. High purity methanol is readily obtained and it is completely miscible with water. Methanol has a high percent carbon content and is relatively cheap, although only a limited number of organisms will metabolize it. Also unlike many other carbon sources, only low concentrations, 0.1-1% (v/v), are tolerated by microorganisms, higher levels being toxic. During fermentations on methanol, the oxygen demand and heat of the fermentations are high, but this is even more problematic when growing on alkanes. Several companies used methanol in microbial protein production in the 1970s and early 1980s, but these processes are currently uneconomic.

Ethanol is less toxic than methanol and is used as a sole or co-substrate by many organisms, but it is too expensive for general use as a carbon source. However, its biotransformation to acetic acid by acetic acid bacteria remains a major fermentation process.

### **Fats and Oils**

Hard animal fats that are mostly composed of glycerides of palmitic and stearic acids are rarely used in fermentations. However, plant oils (primarily from cotton seed, linseed, maize, olive, palm, rape seed and soya) and occasionally fish oil, may be used as the primary or supplementary carbon source, especially in antibiotic production. Plant oils are mostly composed of oleic and linoleic acids, but linseed and soya oil also have a substantial amount of linolenic acid. The oils contain more energy per unit weight than carbohydrates. In addition, the carbohydrates occupy a greater volume because they are usually prepared as aqueous solutions of concentrations no greater than 50% (w/v). Consequently, oils can be particularly useful in fed-batch operations as less spare capacity is needed to accommodate further additions of the carbon source.

#### **2.2.1.2 Nitrogen Sources**

Most industrial microbes can utilize both inorganic and organic nitrogen sources. Inorganic nitrogen may be supplied as ammonium salts, often ammonium sulphate and diammonium hydrogen phosphate, or ammonia. Ammonia can also be used to adjust pH of the fermentation. Organic nitrogen sources include amino acids, proteins and urea. Nitrogen is often supplied in crude forms that are essentially by-products of other industries, such as corn steep liquor, yeast extracts, peptones and soya meal. Purified amino acids are used only in special situations, usually as precursors for specific products.

#### **Corn Steep Liquor**

Corn steep liquor is a by-product of starch extraction from maize and its first use in fermentations was for penicillin production in the 1940s. The extract composition of the liquor varies depending on the quality of the maize and the processing conditions. Concentrated extracts generally contain about 4% (w/v) nitrogen, including a wide range of amino acids, along with vitamins and minerals. Any residual sugars are usually converted to lactic acid (9-20%, w/v) by contaminating bacteria. Corn steep liquor can sometimes be replaced by similar liquors, such as those derived from potato starch production.

#### **Yeast Extracts**

Yeast extracts may be produced from waste baker's and brewer's yeast, or other strains of *S. cerevisiae*. Alternate sources are *Kluyveromyces marxianus* (formerly classified as *K. fragilis*) grown on whey and *Candida utilis* cultivated using ethanol, or wastes from wood and paper processing. Those extracts used in the formulation of fermentation media are normally salt-free concentrates of soluble components of hydrolyzed yeast cells. Yeast extracts with sodium chloride concentrations greater than 0.05% (w/v) cannot be used in fermentation processes due to potential corrosion problems. Yeast cell hydrolysis is often achieved by autolysis using the cell's endogenous enzymes, usually without the need for additional hydrolytic enzymes. Autolysis can be initiated by temperature or osmotic shock, causing cells to die but without inactivating their enzymes. Temperature and pH are controlled throughout to ensure an optimal and standardized autolysis process. Temperature control is particularly important to prevent loss of vitamins. Autolysis is performed at 50-55°C for several hours before the temperature is raised to 75°C to inactivate the enzymes. Finally the cells are disrupted by plasmolysis or mechanical disruption. Cell wall materials and other debris are removed by filtration or centrifugation and the resultant extract is rapidly concentrated. Extracts are



available as liquids containing 50-65% solids, viscous pastes or dry powders. They contain amino acids, peptides, water-soluble vitamins and some glucose, derived from the yeast storage carbohydrates (trehalose and glycogen).

### **Peptones**

Peptones are usually too expensive for large scale industrial fermentations. They are prepared by acid or enzyme hydrolysis of high protein materials: meat, casein, gelatine, keratin, peanuts, soy meal, cotton seeds etc. Their amino acid compositions may vary depending upon the original protein source. For example, gelatine-derived peptones are rich in proline and hydroxyproline, but are almost devoid of sulphur-containing amino acids; whereas keratin peptone is rich in both proline and cystine, but lacks lysine. Peptones from plant sources invariably contain relatively large quantities of carbohydrates.

### **Soya Bean Meal**

Residues remaining after soya beans have been processed to extract the bulk of their oil are composed of 50% protein, 8% non-protein nitrogenous compound, 30% carbohydrates and 1% oil. This residual soya meal is often used in antibiotic fermentations because the components are only slowly metabolized, thereby eliminating the possibility of repression of product formation.

#### **2.2.1.3 Water**

All fermentation processes, except SSF, require vast quantities of water. Not only is water a major component of all media, but it is important for ancillary services like heating, cooling, cleaning and rinsing. A reliable source of large quantities of clean water, of consistent composition, is therefore essential. Important factors to consider when assessing suitability of a water supply are: pH, dissolved salts and effluent contamination. The mineral content is important in brewing (mashing step) and historically influenced the siting of breweries and types of beer produced. Before use, removal of suspended solids, colloids and microorganisms is usually required. When the water supply is "hard", it is treated to remove salts such as calcium carbonate. Iron and chlorine may also require removal. For some fermentations, notably plant and animal cell culture, the water must be highly purified. Water is becoming increasingly expensive, necessitating its recycling/re-usage wherever possible. This minimizes water costs and reduces the volume requiring waste-water treatment.

#### **2.2.1.4 Minerals**

All microorganisms require certain mineral elements for growth and metabolism. In many media, magnesium, phosphorous, potassium, sulphur, calcium and chlorine are essential components and must be added. Others such as cobalt, copper, iron, manganese, molybdenum and zinc are present in sufficient quantities in the water supplies and as impurities in other media ingredients. For example, corn steep liquor contains a wide range of minerals that will usually satisfy the minor and trace mineral needs. Occasionally, levels of calcium, magnesium, phosphorous, potassium, sulphur and chloride ions are too low to fulfil requirements and these may be added as specific salts.

#### **2.2.1.5 Vitamins and Growth Factors**

Many bacteria can synthesize all necessary vitamins from basic elements. For other bacteria, filamentous fungi and yeasts, they must be added as supplements to the fermentation medium. Most natural carbon and nitrogen sources also contain at least some of the required vitamins as minor contaminants. Other necessary growth factors, amino acids, nucleotides, fatty acids and sterols are added either in pure form or for economic reasons, as less expensive plant and animal extracts.

#### **2.2.1.6 Precursors**

Precursors are defined as “substances added prior to or simultaneously with the fermentation which are incorporated without any major change into the molecule of the fermentation product and which generally serve to increase the yield or improve the quality of the product”. They are required in certain industrial fermentations and are provided through crude nutritive constituents, e.g., corn steep liquor or by direct addition of more pure compounds. Some fermentations must be supplemented with specific precursors, notably for secondary metabolite production. When required, they are often added in controlled quantities and in a relatively pure form, examples include, D-threonine is used as a precursor in L-isoleucine production by *Serratia marcesans*, and anthranilic acid additions are made to fermentations of the yeast *Hansenula anomola* during L-tryptophan production. The use of corn steep liquor as side-chain precursors in penicillin fermentations results in six different penicillins as opposed to the use of phenylacetic acid which results in mainly Penicillin G formation.

#### **2.2.1.7 Inducers and Elicitors**

If product formation is dependent upon the presence of a specific inducer compound or a structural analogue, it must be incorporated into the culture medium or added at a specific point during the fermentation. The majority of enzymes of industrial interest are inducible. Inducers are often substrates such as starches or dextrans for amylase. In plant cell culture the production of secondary metabolites, such as flavanoids and terpenoids can be triggered by adding elicitors. These may be isolated from various microorganisms, particularly plant pathogens. Inducers are often necessary in fermentations of genetically modified microorganisms. This is because the growth of genetically modified microorganisms (GMMs) can be impaired when the cloned genes are “switched on”, due to the very high levels of their transcription and translation. Consequently, inducible systems for the cloned genes are incorporated that allow initial maximization of growth to establish high biomass density, whereupon the cloned gene can then be “switched on” by the addition of the specific chemical inducer, e.g., a commercial system developed based on the *alcA* promoter in *Aspergillus nidulans* to express human interferon  $\alpha 2$ . The system is induced by volatile chemicals like ethylmethylketone which are added when the biomass has increased to an adequate level and growth medium contains non-repressing carbon source.

#### **2.2.1.8 Inhibitors**

Inhibitors are used to redirect metabolism towards the target product and reduce formation of other metabolic intermediates; others halt a pathway at a certain point to prevent further metabolism of the target product. An example of an inhibitor specifically employed to redirect

metabolism is sodium bisulphite, which is used in the production of glycerol by *S. cerevisiae*. Some GMMs contain plasmids bearing an antibiotic resistance gene, as well as the heterologous gene(s). The incorporation of this antibiotic into the medium used for the production of the heterologous product selectively inhibits any plasmid-free cells that may arise.

#### **2.2.1.9 Cell Permeability Modifiers**

These compounds increase cell permeability by modifying cell walls and/or membranes, promoting the release of intracellular products into the fermentation medium. Compounds used for this purpose include penicillins and surfactants. They are frequently added to amino acid fermentations, including processes for producing L-glutamic acid using members of the genera *Corynebacterium* and *Brevibacterium*.

#### **2.2.1.10 Oxygen**

Depending on the amount of oxygen required by the organism, it may be supplied in the form of air containing about 21% (v/v) oxygen or occasionally as pure oxygen when requirements are particularly high. The organism's oxygen requirements may vary widely depending upon the carbon source. For most fermentations the air or oxygen supply is filter sterilized prior to being injected into the fermenter. The specific oxygen uptake rate of a microorganism increases with increase in the dissolved oxygen concentration up to a certain point referred to as the critical level ( $C_{crit}$ ). Maximum biomass production is achieved by satisfying the organism's maximum specific oxygen demand by maintaining the dissolved oxygen concentration greater than the critical level. However, frequently the objective of a fermentation is to produce a product and not biomass and the metabolic disturbance of the cell by oxygen starvation may be advantageous to product formation.

#### **2.2.1.11 Antifoams**

Antifoams are necessary to reduce foam formation during fermentation. Foaming is largely due to media proteins that become attached to the air-broth interface where they denature to form a stable foam "skin" that is not easily disrupted. If uncontrolled the foam may block air filters, resulting in the loss of aseptic conditions; the fermenter becomes contaminated and microorganisms are released into the environment. Of possibly the most importance is the need to allow "freeboard" in fermenters to provide space for the foam generated. If foaming is minimized, then throughputs can be increased.

There are three possible approaches to controlling foam production: the use of a defined medium and a modification of some of the physical parameters, e.g. pH, temperature, aeration and agitation (if the foam is due to media components), use of chemical foam breakers and addition of chemical antifoams. Antifoams are surface active agents that reduce the surface tension in the foams and destabilize the protein films by (i) hydrophobic bridges between two surfaces; (ii) displacement of the adsorbed protein; and (iii) rapid spreading on the surface of the film. The ideal antifoam should have the following properties:

- 1 readily and rapidly dispersed with rapid action
- 2 high activity at low concentrations
- 3 prolonged action

- 4 non-toxic to fermentation microorganisms, humans or animals
- 5 low cost
- 6 thermostability
- 7 compatibility with other media components and the process, i.e., having no effect on oxygen transfer rates or downstream processing operations
- 8 be heat sterilisable

Natural antifoams include plant oils (e.g., from soya, sunflower and rapeseed), deodorized fish oil, mineral oils and tallow. The synthetic antifoams are mostly silicon oils, poly alcohols and alkylated glycols. Since antifoams are of low solubility, they need a carrier, e.g., lard oil, liquid paraffin or castor oil, which may be metabolised and therefore affect the fermentation process. Many of the surface-active agents, particularly the oils, are added as emulsions of suspended oil droplets which can destabilise the foams by acting as hydrophobic bridges between the two film surfaces or by displacing the stabilising adsorbed material, e.g. protein, at the bubble–liquid interface. However, those conditions, which cause collapse of the foam structure, can also favour the coalescence of bubbles in the body of the liquid. This results in an increase in the mean bubble diameter and a reduction in gas hold-up. Both of these effects will tend to reduce the specific interfacial area available for mass transfer. The concentrations of many antifoams which are necessary to control foaming may reduce the oxygen transfer rate by as much as 50%. Thus, antifoam addition should be kept to an absolute minimum. Some antifoams may reduce the oxygen transfer rate as well as adversely affect downstream processing steps, especially membrane filtration. If the oxygen transfer rate is too severely affected mechanical foam breakers may have to be considered.

#### 2.2.2 ANIMAL CELL CULTURE MEDIA

Animal cell culture media are normally based on complex basal media, such as Eagle's cell culture medium, which contains glucose, mineral salts, vitamins and amino acids. For mammalian cells a serum is usually added, such as foetal calf serum, calf serum, newborn calf serum or horse serum. Sera provide a source of essential growth factors, including initiation and attachment factors and binding proteins. They also supply hormones, trace elements and protease inhibitors.

The highly complex composition of sera makes substitution with lower cost ingredients very difficult. Sterilization of formulated animal culture media and media constituents is also more problematic as many components are thermolabile, requiring filter sterilization. Normally, sera constitute 5-10% (v/v) of the medium, but attempts have been made to reduce and ultimately eliminate its use. This is necessary due to its high cost and the fact that it is a potential source of prions and viruses. In some circumstances levels have now been lowered to 1-2% (v/v) and some cell lines have been developed that grow in serum-free media.

#### 2.2.3 PLANT CELL CULTURE MEDIA

In contrast to animal cell culture media, those used for plant cell culture are usually chemically defined. They contain an organic carbon source (as most plant cells are grown heterotrophically), a nitrogen source, mineral salts and growth hormones. Sucrose is

frequently incorporated as the carbon source, particularly for secondary metabolite production, but glucose, fructose, maltose and even lactose have been used. Nitrate is the usual nitrogen source, often supplemented with ammonium salts. However, some species may require organic nitrogen, normally in the form of amino acids. The combination and concentration of plant hormones provided depend upon the specific fermentation. Auxins are usually supplied, along with cytokinins to promote cell division. A two-phase culture has often proved to be useful in increasing productivity, particularly for producing secondary metabolites such as shikonin. The first phase uses a medium optimized for growth; the second phase promotes product formation